

Theoretical Evidence for Rotational Energy Cluster Formation in the Vibrational Ground State of SO₃

Daniel S. Underwood^a, Sergei N. Yurchenko^a, Jonathan Tennyson^a, Per Jensen^b

^aDepartment of Physics and Astronomy, University College London,
Gower Street, London WC1E 6BT, ^aE-mail: daniel.underwood@ucl.ac.uk

^bTheoretische Chemie, Bergische Universität, D-42097 Wuppertal, Germany

In previous work [1] we have computed energy levels and transition intensities for SO₃ adequate for simulating a room-temperature spectrum. This involved the calculation of ro-vibrational energy levels and associated eigenfunctions, for rotational states up to and including $J = 85$, using the TROVE program [2]. The theoretical microwave spectrum composed from pure-rotational transitions exhibits some interesting features indicating the presence of localized stable rotation axes at high J . In this work we aim to investigate the rotational dynamics of SO₃ using our theoretical approach. Following a previous, analogous study of PH₃ [3], we compute the rotational energy levels in the vibrational ground state of SO₃ up to a high value of rotational quantum number $J (> 200)$. At such high values of J we observe a clustering of rotational energy levels with a six-fold pattern similar to that observed for PH₃. To our knowledge this is the first time such effects have been observed for a trigonal D_{3h} planar molecule.

We also present some details of our efforts to extend upon our initial room-temperature line-list, including a spectroscopic refinement of our previous *ab initio* potential energy surface and some computational difficulties we have observed in attempting to prepare for a hot line-list calculation.

[1] D. S. Underwood, S. N. Yurchenko, J. Tennyson, Phys. Chem. Chem. Phys., **2013**, *15*, 10118-10125

[2] S. N. Yurchenko, W. Thiel, P. Jensen, J. Mol. Spec., **2007**, *245*, 126-140

[3] S. N. Yurchenko, W. Thiel, S. Patchkovskii, P. Jensen, Phys. Chem. Chem. Phys., **2005**, *7*, 573-582

[1] D. S. Underwood, S. N. Yurchenko, J. Tennyson, Phys. Chem. Chem. Phys., **2013**, *15*, 10118-10125

[2] S. N. Yurchenko, W. Thiel, P. Jensen, J. Mol. Spec., **2007**, *245*, 126-140

[3] S. N. Yurchenko, W. Thiel, S. Patchkovskii, P. Jensen, Phys. Chem. Chem. Phys., **2005**, *7*, 573-582